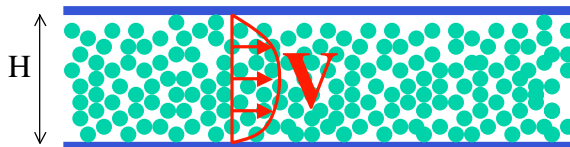
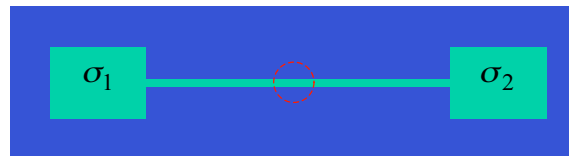


# Coupling Stokes's creep and Herring's diffusion at a small length scale

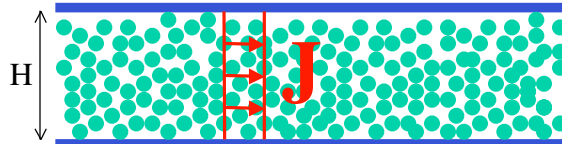
Zhigang Suo  
*Harvard University*

International Workshop on Nanomechanics  
Asilomar Conference Grounds  
14-17 July 2004

## Stress gradient drives mass transport



$$q_{\text{creep}} = \frac{H^3}{12\eta} \nabla \sigma$$



$$q_{\text{diffuse}} = \frac{HD\Omega}{kT} \nabla \sigma$$

When does transition occur?

	Stokes's creep	Herring's diffusion
Kinematics	$d_{ij} = \frac{1}{2}(v_{i,j} + v_{j,i}) \quad v_{k,k} = 0$	$J_{k,k} = 0$
Kinetics	$\sigma_{ij} = s_{ij} + \sigma\delta_{ij}, \quad s_{ij} = 2\eta d_{ij}$	$J_k = -\frac{D}{kT\Omega} \mu_{,k}$
Force balance	$\sigma_{ij,j} = 0$	?
Boundary conditions	Prescribe traction or velocity	Prescribe chemical potential $\mu = -\Omega\sigma_n$

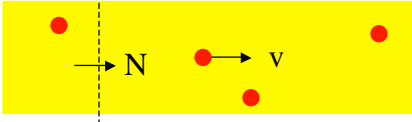
How to couple creep and self-diffusion?

Suo, Kubair, Evans, Clarke, Tolpygo  
Acta Materialia 51, 959 (2003)

## Kinematics

$$N_k = J_k + v_k / \Omega$$

↑ net flux
↑ diffusion flux
↑ convection

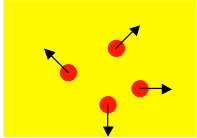


marker velocity

Net flux is divergence-free  $J_{k,k} + v_{k,k} / \Omega = 0$


Strain-rate measures relative marker motion

$$d_{ij} = \frac{1}{2}(v_{i,j} + v_{j,i})$$



Diffusion-induced strain-rate (isotropic placement rule)

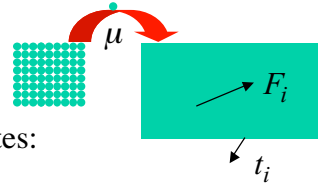
$$d_{ij}^D = -\frac{\Omega}{3} J_{k,k} \delta_{ij}$$



Creep strain-rate

$$d_{ij}^C = \frac{1}{2}(v_{i,j} + v_{j,i}) - \frac{1}{3} v_{k,k} \delta_{ij}$$

# Energetics



Define driving forces as energy-conjugates:

$$\int (s_{ij} d_{ij}^C + f_i J_i) dV + \int \sigma (v_{k,k} + \Omega J_{k,k}) dV = \int F_i J_i dV + \int t_i v_i dA - \int \mu J_i n_i dA$$

power dissipation
kinematic constraint
power supply

$$-\int (s_{ij} + \sigma \delta_{ij})_{,j} v_i dV + \int ((s_{ij} + \sigma \delta_{ij})_{,j} n_j - t_i) J_i dA + \int (f_i - F_i - \Omega \sigma_{,i}) J_i dV + \int (\Omega \sigma + \mu) J_i n_i dA = 0$$

$(s_{ij} + \sigma \delta_{ij})_{,j} = 0$	$(s_{ij} + \sigma \delta_{ij})_{,j} n_j = t_i$
$f_i = F_i + \Omega \sigma_{,i}$	$\mu = -\Omega \sigma$

Suo, JAM, in press

# Kinetics

diffusion  $J_i = \frac{Df_i}{kT}$

creep  $d_{ij}^C = \frac{s_{ij}}{2\eta}$

## Summary of equations for $\mathbf{v}$ and $\sigma$

$$\left[ \eta \left( v_{i,j} + v_{j,i} - \frac{2}{3} v_{k,k} \delta_{ij} \right) \right]_{,j} + \sigma_{,i} = 0$$

$$v_{k,k} = - \left( \frac{D}{kT} (F_k + \Omega \sigma_{,k}) \right)_{,k}$$

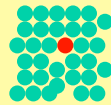
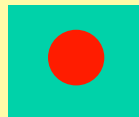
Characteristic length  $\Lambda = \sqrt{\frac{\eta D \Omega}{kT}}$

Suo, JAM, in press

## Mechanistic pictures: self-diffusion and creep

$$\Lambda = \sqrt{\frac{\eta D \Omega}{kT}}$$

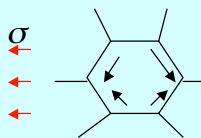
Stokes-Einstein



$$D = \frac{kT}{6\pi a \eta}$$

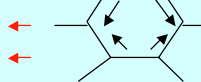
$$\Lambda \sim a$$

Nabarro-Herring



$$\frac{1}{\eta} \sim \frac{D \Omega}{kT d_g^2}$$

Coble



$$\frac{1}{\eta} \sim \frac{D_b \delta_b \Omega}{kT d_g^3}$$

$$\Lambda \sim d_g$$

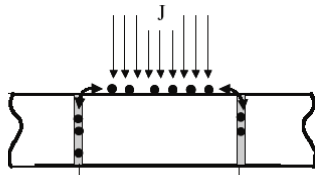
Needleman-Rice

Creep in grains,  
Diffusion on grain boundaries

$$\Lambda = \sqrt{\frac{\eta_* D \Omega}{kT}}$$

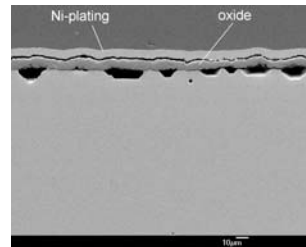
# Injecting and emitting atoms

Deposition-induced compression  
(chemical potential excess)



Chason et al. PRL 88, 156103 (2002)

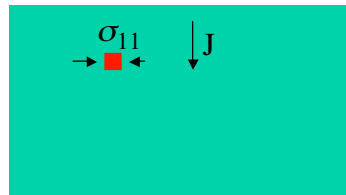
Oxidation-induced tension: NiAl  
(chemical potential deficit)



Suo, Kubair, Evans, Clarke, Tolpygo  
Acta Materialia 51, 959 (2003)

# Thick substrate: stress generation

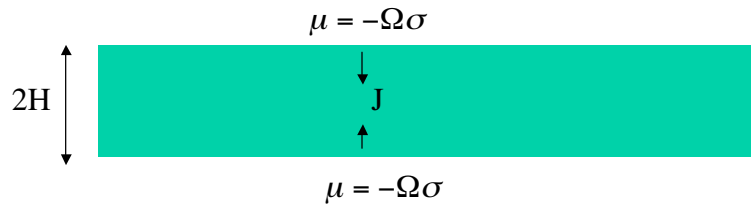
$$\mu = -\Omega\sigma$$



$$\sigma_{11}(x_3) = -\frac{3\mu}{2\Omega} \exp(x_3/l)$$

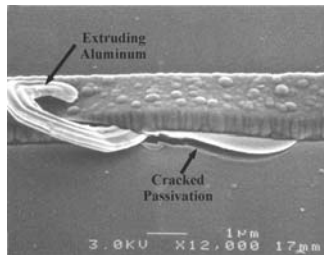
$$l = \sqrt{\frac{4D\eta\Omega}{3kT}}$$

## Thin foil lateral expansion/contraction

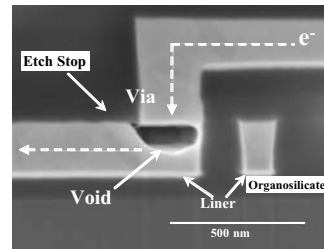


$$d_{11} = \left( \frac{\mu}{4\eta\Omega} \right) \frac{\sinh(H/l)}{(H/l)\cosh(H/l) - \sinh(H/l)}$$

## Electromigration

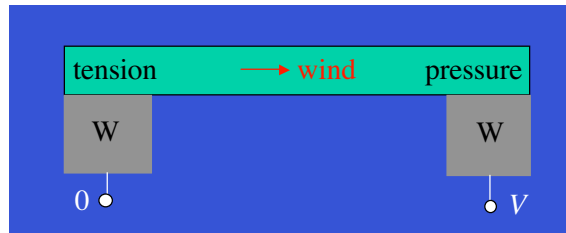


Chiras, Clarke  
JAM 88, 6302 (2000)



He, Suo, Marieb, Maiz  
Submitted

## Electron wind vs. stress gradient

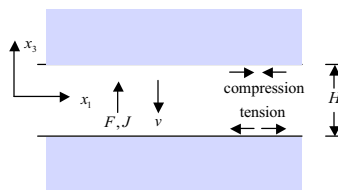


Electron wind drives diffusion  $\frac{HDF}{kT\Omega}$

Stress gradient drives diffusion and creep  $\frac{HD}{kT}\nabla\sigma + \frac{H^3}{12\eta}\nabla\sigma$

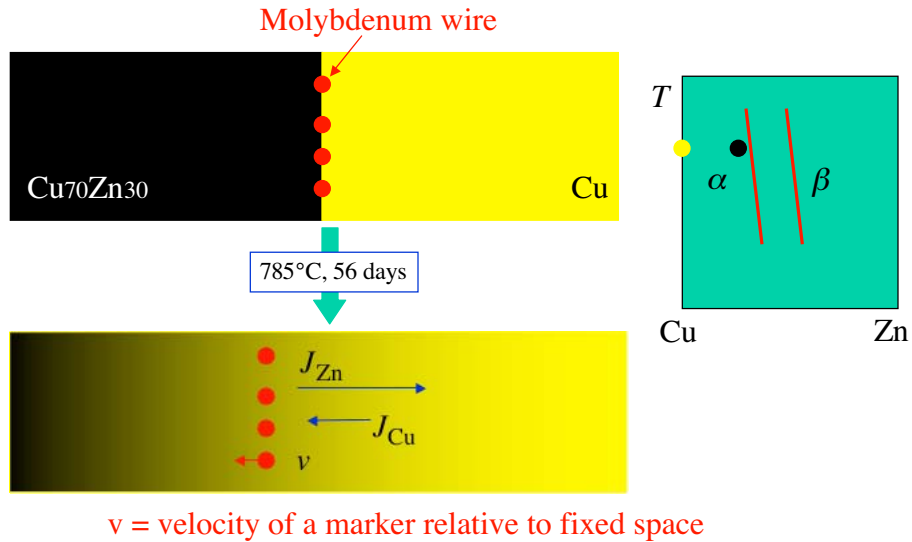
No net flow if  $F = \left(1 + \frac{H^2}{12\Lambda^2}\right)\Omega\nabla\sigma$  (Blech condition generalized)

## Electromigration through a foil



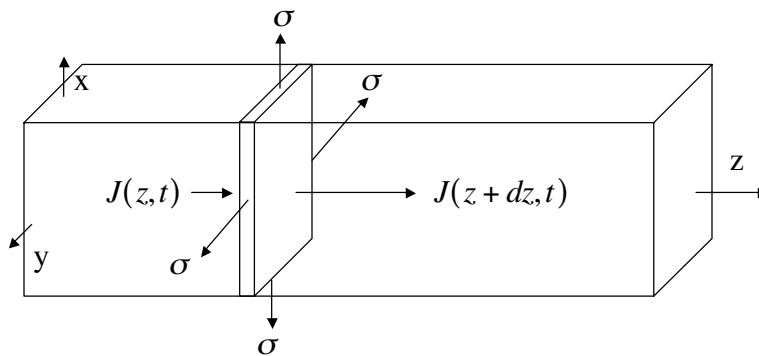
$$\sigma_{11} = -\frac{3Fl \sinh(x_3/l)}{2\Omega \cosh(H/2l)}$$

## Nonreciprocal diffusion The Kirkendall Effect



Smigieskas and Kirkendall, Trans. Am. Inst. Min. Engrs. **171**, 130 (1947)

## Transverse constraint



## Co-evolution of composition and stress

Number conservation

$$\frac{\partial C^A}{\partial t} + (J_i^A + C^A v_i)_{,i} = 0$$

diffusion kinetics

$$J_i^A = D^A \left[ -\frac{\partial C^A}{\partial x_i} + \frac{c}{\phi kT} \frac{\partial \sigma_m}{\partial x_i} \right]$$

force balance

$$\sigma_{ij,j} = 0$$

mean stress

$$\sigma_m = \sigma_{kk} / 3$$

deviatoric stress

$$s_{ij} = \sigma_{ij} - \sigma_m \delta_{ij}$$

effective shear stress

$$\tau_e = \sqrt{s_{ij}s_{ij} / 2}$$

$$d_{ij} = d_{ij}^C + d_{ij}^D$$

total strain - rate

$$d_{ij} = \frac{1}{2} (v_{i,j} + v_{j,i})$$

creep strain - rate

$$d_{ij}^C = f(\tau_e) \frac{s_{ij}}{2\tau_e}$$

dilatation strain - rate

$$d_{ij}^D = -\frac{\Omega}{3} (J_k^A + J_k^B)_{,k} \delta_{ij}$$

## Summary

- Diffusion and creep are coupled
- Generate stress by
  - excess chemical potential
  - electron wind
  - nonreciprocal diffusion

[www.deas.harvard.edu/suo](http://www.deas.harvard.edu/suo)  
Publication 135, 156